

### Silicon Photocathode with Vertically Aligned TiO<sub>2</sub> Nanorods and Ni Catalyst for Enhancing Photoelectrochemical Nitrate Reduction to Ammonia



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# 2.Methods

# **3.Results**

# 4.Summary



GIST



✓ Essential compound in industry

& agriculture

✓ Consumption of natural gas



#### ✤ Eco-friendly catalytic ammonia synthesis



the waste nitrate

GIS

✓ High ammonia yield & ease of operation

#### However, e-NO<sub>3</sub>RR requires a high potential.





The low yield rate & selectivity remain bottlenecks.

< 4/14 >

#### Silicon (Si) photoelectrodes

![](_page_4_Picture_3.jpeg)

#### Advantage

- ✓ Narrow band gap (1.12 eV)
  - → High photocurrent density
- Excellent charge carrier mobility
- ✓ Mature fabrication

#### **Drawbacks**

- Insufficient photovoltage
- Chemical corrosion in aqueous electrolytes
- Sluggish charge-transfer kinetics
- High light reflection

#### So, passivation layer & Catalysts are required.

![](_page_4_Picture_15.jpeg)

#### Titanium dioxide (TiO<sub>2</sub>)

![](_page_5_Picture_3.jpeg)

- ✓ Ultra-Violet light absorption capability
- ✓ Efficient anti-reflection
- ✓ Appropriate band offset with silicon

#### Nickel (Ni) catalyst

![](_page_5_Figure_8.jpeg)

![](_page_5_Picture_9.jpeg)

- ✓ Cost effectiveness
- ✓ Stable in alkaline electrolyte
- ✓ Reaction kinetic improvement

#### TiO<sub>2</sub> nanorods + Ni catalyst

"Enhancing PEC NO<sub>3</sub>-RR"

![](_page_5_Picture_15.jpeg)

# **2.** Methods

3. Ni decoration

- ✤ Sequential deposition by e-beam evaporator
- 1. TiO<sub>2</sub> film (10 nm) 2. TiO<sub>2</sub> n

![](_page_6_Figure_4.jpeg)

![](_page_6_Figure_5.jpeg)

![](_page_6_Picture_6.jpeg)

Surface morphology analysis (Scanning Electron Microscope (SEM) images)

![](_page_7_Figure_3.jpeg)

1: Bare p-Si 2: TiO<sub>2</sub> NRs/ TiO<sub>2</sub> film/ p-Si

![](_page_7_Picture_5.jpeg)

 $\checkmark$  The vertically grown TiO<sub>2</sub> NRs exhibited uniform length and porosity.

![](_page_7_Picture_7.jpeg)

Structural characterizations (X-ray photoelectron Spectroscopy (XPS))

![](_page_8_Figure_3.jpeg)

- $\checkmark$  Ni catalysts is embedded in the TiO<sub>2</sub> NRs.
- ✓ The abundant surface oxygen vacancies can further enhance the catalytic activity.

![](_page_8_Picture_6.jpeg)

✤ PEC performances for NO<sub>3</sub>-RR (TiO<sub>2</sub> NRs/ TiO<sub>2</sub> film/ p-Si photocathode)

![](_page_9_Figure_3.jpeg)

 $\checkmark$  The Si photocathodes with TiO<sub>2</sub> NRs exhibited good PEC performance for PEC NO<sub>3</sub>-RR.

![](_page_9_Picture_5.jpeg)

✤ PEC performances for NO<sub>3</sub>-RR (Ni/ TiO<sub>2</sub> NRs/ TiO<sub>2</sub> film/ p-Si photocathode)

![](_page_10_Figure_3.jpeg)

 $\checkmark$  Ni catalysts enhanced PEC NO<sub>3</sub>-RR with an onset potential of 0.28 V<sub>RHE</sub>.

![](_page_10_Picture_5.jpeg)

✤ Quantitative analyzation for PEC NO<sub>3</sub>-RR

![](_page_11_Figure_3.jpeg)

 $\checkmark$  At -0.2 V<sub>RHE</sub>, a Faradaic efficiency of 90.6% and a yield of 1046.6652 µg/h cm<sup>2</sup> were achieved.

![](_page_11_Picture_5.jpeg)

- Using GLAD method by e-beam evaporator, we fabricated silicon photocathode with vertically aligned TiO<sub>2</sub> nanorod arrays and nickel.
- ✓ The decorated Ni on vertically aligned TiO<sub>2</sub> nanorod arrays completely increased the reaction surface area and the enhanced photo-absorption was enabled by the anti-reflection of the 1D nanorod structure.
- ✓ It showed photocurrent of 31.8 mA cm<sup>-2</sup> at -0.2 V<sub>RHE</sub> and positive onset potential of 0.28 V<sub>RHE</sub> and 1046 ug/h cm<sup>2</sup> ammonia yield.

![](_page_12_Picture_6.jpeg)

![](_page_12_Figure_7.jpeg)

![](_page_12_Picture_8.jpeg)

4. Summary

# Thank you for your attention

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![](_page_13_Picture_2.jpeg)

![](_page_13_Picture_3.jpeg)